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1. Generally non-polar (won't dissolve in water) 2. Soluble in non-polar solvents 3. Non-electrolytes. 4. Low melting points ORGANIC CHEMISTRY Continuation of Organic Chemistry I. Methods of analysis,

Chemistry Hydrocarbons Study Guide Answers

Hydrocarbons Alkanes Study Guide Answers Alkanes have only a single bond in its molecular formula. A hydrocarbon molecule is said to be saturated if there are no double or triple bonds. Alkenes have one or more than one double bond. Answer true or false: Alkanes are saturated

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hydrocarbons ...

Hydrocarbons Alkanes Study Guide Answers

ChemFacts. • The primary source of hydrocarbons is petroleum. • About 75 million barrels of petroleum are pumped out of the Earth each day. • Hydrocarbons are used as fuels and are the raw materials for products such as plastics, synthetic fibers, solvents, and industrial chemicals.

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organic molecules that only
contain two elements: carbon and
hydrogen. Common hydrocarbons
include methane, propane, and
butane. Section 22hydrocarbon
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Guide Key Concepts 22.1

Hydrocarbons • Because carbon
has four valence electrons, car-
bon atoms always form four
covalent bonds. • The carbon
atoms in an alkane can be
arranged in a straight chain or in
a chain that has branches.

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- The primary source of hydrocarbons is petroleum.
- About 75 million barrels of petroleum are pumped out of the Earth each day.
- Hydrocarbons are used as fuels and are the raw materials for products such as plastics, synthetic fibers, solvents, and industrial chemicals.

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Hydrocarbons Answer

Chapter 22 Hydrocarbons Study Guide Answers Hydrocarbons are organic molecules that only contain two elements: carbon and hydrogen. Common hydrocarbons

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Answers
include methane, propane, and butane. Hydrocarbons are the primary constituents of fossil...

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controlled process by which
hydrocarbons are broken down or
rearranged into smaller, more
useful molecules natural gas This
is typically made up of 80%

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methane, 10% ethane, Page 5/29

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of carbon and its compounds.

simplest organic compound,
composed of hydrogen and
carbon. hydrocarbons with single
bonds, "ane" ending.

hydrocarbons with double bonds,
"ene" ending. Organic Chemistry.

the study of carbon and its
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Answers chains or rings. •
Common Characteristics:

1. Generally non-polar (won't
dissolve in water) 2. Soluble in non-
polar solvents 3. Non-electrolytes.

4. Low melting points ORGANIC

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analysis, chemistry of

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Hydrocarbons An infinite variety
of compounds can be assembled
from only carbon and hydrogen
atoms. Such hydrocarbons are
the simplest organic compounds,
but they are also of prime

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economic importance because they include the constituents of petroleum and natural gas.

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