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4.2.2 How much energy is released when bonds form to produce 2 mol of H₂O and 1 mol of CO₂? (4) 4.2.3 Hence calculate the energy transferred (?H) for the combustion of 1 mol CH₄. (2) 4.2.4 Draw a labelled energy profile for this reaction. Include the labels: ENTHALPY CHANGE (?H) ACTIVATION ENERGY (E_a) (6)

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